

RVS COLLEGE OF ARTS AND SCIENCE

Autonomous and Affiliated to Bharathiar University, Approved by AICTE

Re Accredited with 'A+' Grade by NAAC

Sulur, Coimbatore – 641 402.



DEPARTMENT OF BIOCHEMISTRY

M.Sc. BIOCHEMISTRY

(Effective from the Academic Year 2023- 2024)

HOD

PRINCIPAL

COE

VISION

Quality Education for Digital Era.

MISSION

To impart a need – based quality education through comprehensive curriculum by adopting apt technologies and progressive teaching, learning and research processes.

ABOUT THE DEPARTMENT

The Department of Biochemistry was established in 1996. The department is functioning with under graduate, Post graduate and Research programs with a team of 11 faculty members.

The Biochemistry focuses on life processes at the molecular level, that emphasis on the biochemistry and molecular biology of cellular metabolism and Bioenergetics.

The curriculum is designed for students seeking preparation for graduate studies, for the health-care professions, or for entry-level positions in science-related industries. The teaching and learning process is well executed properly and reveals with sophisticated instruments.

EXECUTIVE MEMBERS

CHAIRMAN

Dr. K.V. Kupusamy

MANAGING TRUSTEE

Dr. K. Senthil Ganesh

CORRESPONDENT

Mrs. S. Srividyalakshmi Senthil Ganesh

SECRETARY

Prof. Saramma Samuel

PRINCIPAL

Dr. T. Sivakumar

VICE PRINCIPAL

Dr. M.P. Ayyappadas

CONTROLLER OF EXAMIATIONS

Ms. G. Jeyalakshmi

HEAD OF THE DEPARTMENT

Dr. S.Shamina

Department: BIOCHEMISTRY

Programme: M.Sc., BIOCHEMISTRY

Programme code: 1PBC

PROGRAMME OUTCOMES (POs)

PO1	Domain Knowledge
PO2	Communicative Competence
PO3	Digital Strategic Knowledge
PO4	Multi-Cultural Competence
PO5	Critical Thinking and Problem Solving
PO6	Research and Analytical Skills
PO7	Moral, Ethical and Professional Responsibilities
PO8	Leadership and Lifelong Learning

PROGRAMME SPECIFIC OUTCOMES: (PSOs)

Upon completion of Master of Biochemistry, STUDENTS are able to achieve the following outcomes.

PSO1	Assess the fundamental concepts of experimental and theoretical techniques in Molecular Gene expression, Proteomics, Pathology, Clinical research and Nano-sciences.
PSO2	Demonstrate and operate the technologies and thereby gaining insights from clinical data to make cognizant predictions.
PSO3	Implement research oriented learning to develop methodical and integrative problem-solving approaches in the biochemical industries and to validate effective projects for lifelong learning.
PSO4	Construct and formulate pre-clinical studies to invent innovative ideas to develop their employability and entrepreneurial skills.

RVS COLLEGE OF ARTS AND SCIENCE (Autonomous)
Sulur, Coimbatore – 641 402
SCHEME OF EXAMINATIONS
M.Sc. BIOCHEMISTRY
2023- 2025 BATCH

Semester	Course Opted	Course Name	D	L	T	P	CIA	SEE	Marks	Credits
SEMESTER I	M-I	Biochemical Techniques	3	4	-		25	75	100	4
	M-II	Human Cell Molecular Biology	3	4	-		25	75	100	4
	M-III	Enzymes and Enzyme Technology	3	4	-		25	75	100	4
	M-IV	Human physiology, Hormones and signal transduction	3	4	-		25	75	100	4
	MP-I	Lab in Computational and Molecular modelling	3	-	-	4	25	75	100	4
	MP-II	Lab in Quantitative analysis of Biological Compounds	-	-	-	5	-	-	-	-
	MP-III	Lab in Enzymology, Plant Biochemistry and Plant Tissue Culture	3	-	-	5	-	-	-	-
	ES	Employability skills	-	2	-	-	-	-	-	-
	Total				32				500	20

Semester	Course Opted	Course Name	D	L	T	P	CIA	SEE	Marks	Credits
SEMESTER II	M-V	Metabolic Regulation	3	4	-	-	25	75	100	4
	M-VI	Advanced Clinical Biochemistry	3	4	-	-	25	75	100	4

	M-VII	Plant Biochemistry and phytotherapeutics	3	4	-	-	25	75	100	4
	M-VIII	Stem cells and Cancer Biology	3	4	-	-	25	75	100	4
	MP-II	Lab in Quantitative analysis of Biological Compounds	6	-	-	5	25	75	100	4
	MP-III	Lab in Enzymology, Plant Biochemistry and Plant Tissue Culture	6	-	-	5	25	75	100	4
	EL-I	Elective - I	3	4	-	-	25	75	100	4
	ES	Employability skills	-	2	-	-	-	-	-	-
	Total			32					700	28

Semester	Course Opted	Course Name	D	L	T	P	CI A	SEE	Marks	Credits
SEMESTER III	M- IX	Pharmaceutical Biochemistry	3	4	-		25	75	100	4
	M-X	Industrial Biochemistry	3	4	-		25	75	100	4
	M-XI	Immunology and Immuno Techniques	3	4	-		25	75	100	4
	M-XII	Clinical Research and Clinical Data analytics	3	4	-		25	75	100	4
	MP-IV	Lab in Immunology and Recombinant DNA technology	-	-	-	5	-	-	-	-
	MP-V	Lab in Research Techniques in Biochemistry	-	-	-	5	-	-	-	-
	EL-II	Elective II- Internship / Mini Project <i>Viva Voce</i>	-	-	-	-	25	75	100	5

EL-III	Elective III- EDC	3	4	-	-	25	75	100	4
ES	Employability skills		2					GRADE	2*
Total			32					600	25

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Semester	Course Opted	Course Name	D	L	T	CI A	SE E	Marks	Credits	
SEMESTER IV	M- XIII	Biostatistics and Research Methodology	3	4	-		25	75	100	4
	MP-IV	Lab in Immunology and Recombinant DNA technology	6	-	-	5	25	75	100	4
	MP-V	Lab in Research Techniques in Biochemistry	6	-	-	5	25	75	100	4
	PV	Major Project Report & Viva Voce	-	-	-		25	75	100	5
	Total				14				400	17
TOTAL								2200	90+2*	

M - Major Paper

MP- Major Practical

EL – Elective

EDC - Extra Disciplinary Course

PV- Major Project Report & Viva Voce

ELECTIVE – I : 1. Recombinant DNA technology 2. Bioinformatics

ELECTIVE – II : 1. Internship 2. Mini Project Viva Voce

ELECTIVE – III : Extra Disciplinary Course (EDC)- List enclosed

Employability Skills (ES) classes are held in first three semesters.

Exams will be conducted at the end of the III semesters.wc .

Two Extra Credits will be given. This is mandatory to get a degree.

ELECTIVE III COURSES (EL III) : (III-SEMESTER)										
ELECTIVE –III-EXTRA DISCIPLINARY COURSE:(EDC)										
Course	CourseName	L	T	P	CIA	ESE	Total	Credits		
One Course–From the Group										
M.A.English	English for Technical Writing	4	-	-	25	75	100	4		
M.Sc Maths	Quantitative Aptitude	4	-	-	25	75	100	4		
M.Sc Applied Electronics	Mobile Communication	4	-	-	25	75	100	4		
M.Sc Food and Nutrition	Food Preservation and Quality Control	4	-	-	25	75	100	4		
M.Sc Biochemistry	Hospital Management	4	-	-	25	75	100	4		
M.Sc Biotechnology	Forensic Science	4	-	-	25	75	100	4		
M.COM(IB) (International Business)	Global Business	4	-	-	25	75	100	4		
M.Sc Microbiology	Diagnostic Microbiology	4	-	-	25	75	100	4		
Total							100	4		

Course Title : BIOCHEMICAL TECHNIQUES (T)	Course Code : 13A
Semester : I	Course Group : M- I
Teaching Scheme in Hrs (L:T:P) : (4:0:0)	Credits : 4 Credits
Map Code: -	Total Contact Hours: 60
CIA: 25 Marks	SEE # : 75 Marks
Programme: MSC-BIOCHEMISTRY	# - Semester End Exam

No	Course Outcome (Cos): After completion of this course, the students will be able to	PSOs	CL.ses.	BLOOM'S TAXONOMY LEVEL
CO1	Demonstrate and illustrate working principle and instrumentation of pH meter and centrifugation technique	PS01	12	AP
CO2	Detect and distinguish chromatographic technique and its application.	PS02	12	AN
CO3	Apply and the role of electrophoresis techniques in separation of molecules.	PS03	12	AP
CO4	Assess the spectroscopic techniques and its application.	PS02	12	E
CO5	Categorize the theoretical principles of radioactivity in clinical aspects.	PS04	12	AN

UNIT-I (LECTURE HOURS-12)

BUFFERS

Introduction-(Basics of acidity & basicity)

pH –(Definition,determination of pH using indicators

HendersonHasselbach equation - Relationship between pKa& Ph)

pH meter – (Working principle and instrumentation)

Types of electrodes – (Principle & working of calomel,silver/silver chloride, hydrogen electrodes)

Centrifugation

(Introduction - Basic principles of centrifugation-Relative centrifugal force)

Preparative and Analytical centrifugation – (Differential centrifugation, density gradient centrifugation,

Basic principlesof sedimentation

Ultracentrifuge - Rotors-types,analytical ultra centrifuge,applications)

Activity

- Differentiate the solutions given based on its pH using pH meter and pH paper .
- Group discussion on different types of centrifugation

UNIT-II (LECTURE HOURS-12)

CHROMATOGRAPHY

Introduction – (Definition, stationary phase, mobile phase, retardation factor)

Paper chromatography. – (Ascending, descending, Circular -Principle, Technique & applications)

Thin layer chromatography – (Adsorbents, solvents used, Principle, Technique & applications)

Ion exchange chromatography - Ion exchangers- anion & cation exchange resins. Principle, Technique & applications)

Molecular sieve chromatography – (Types of gels, Principle, technique & applications, Molecular weight determination of proteins.

Affinity chromatography - Types of ligands, Principle, Technique & applications)

GLC – (Carrier gas, columns, Principle, Technique & applications.

HPLC-Pumping systems, column packing Principle, Technique & applications)

HPTLC- (Principle, Technique & applications)

Activity

- Packing of column using burette, cotton, sand and silica gel -
- Quiz on Chromatography techniques using Kahoot app

UNIT -III (LECTURE HOURS-12)

ELECTROPHORESIS

Introduction – (Definition, migration of ions in electric field, factors affecting electrophoretic mobility.

Paper electrophoresis - Types (Horizontal, vertical) Principle, Technique & applications.

Gel electrophoresis and Immuno electrophoresis - Types of gel electrophoresis-Agarose, SDS –PAGE)

2D Gel electrophoresis.

Pulse field Gel electrophoresis – (Principle, Technique & applications)

Immuno electrophoresis-Single radial immunodiffusion, Ouchterlony double diffusion, Rocket immune electrophoresis.

Isoelectric focusing - Ampholytes Principle, technique & applications)

Blotting

(Southern blot – Principle, Techniques for DNA detection)

Northern and western blot – Principle, Techniques for RNA and protein detection).

Activity

- Chart work to differentiate Horizontal electrophoresis and Vertical electrophoresis.
- Using paper model for blotting techniques- Capillary, electro transfer and semi-dry blotting.

UNIT-IV (LECTURE HOURS-12)

SPECTROMETRY

Introduction - Basic principles, laws of absorption, Beer Lamberts law.

Colorimetry – Principle, Instrumentation & applications

UV-Visible spectrophotometry – Principle, Instrumentation & applications

Flame spectrophotometry - Emission flame & Atomic absorption spectrophotometry

Advanced Techniques – ICP-MS, ICP-OES (Instrumentation & applications.)

Spectrofluorimetry–(Fluorescence, phosphorescence) Techniques, Instrumentation & applications.

FTIR (Fourier Transform Infrared Spectroscopy) – Principle, Instrumentation & applications.

Activity

- Demonstration on principle and working of colorimeter (selection of filter using coloured solution).
- Interpretation on the reports submission of UV- VIS/ FTIR.

UNIT -V (LECTURE HOURS-12)

ISOTOPES

Introduction – (Definition, radioactive decay-negatron, positron, alpha emission, electron capture, gamma rays).

Radioactivity

(Units of radioactivity - Curie, Bequerel, specific activity).

Measurement of radioactivity methods

(Gas ionization - GM counter, proportion counter

Excitation - Scintillation counting)

Photographic method – (Autoradiography)

Clinical applications – (Tracer techniques, Diagnosis, Therapy).

Activity

- Collect related literatures on environmental disasters of radioactive isotopes and methods of disposal of radioactive waste .
- Prepare a chart to categorize the isotopes used for cancer studies.

Text Books :

T1-Analytical Techniques in Biochemistry and Molecular Biology - Authors: Katoch, Rajan, 2011.

T2- Principles and Techniques of Biochemistry and Molecular biology – Edition 7th - Keith Wilson and John Walker, 2016.

T3 - An introduction to Practical Biochemistry- Edition 3rd - David Plummer- 2017.

Reference Books:

R1 –Biophysical chemistry, Avinash Upadhyay and Kakoli Upadhyay, 4th Edition 2019.

R2- Introduction to Principles and Techniques of Biochemistry and Molecular Biology , Wilson and Walker's -Edition 8th 2018

Course Title : HUMAN CELL MOLECULAR BIOLOGY (T)	Course Code : 13B
Semester : I	Course Group : M- II
Teaching Scheme in Hrs (L:T:P) : (4:0:0)	Credits : 4 Credits
Map Code: -	Total Contact Hours: 60
CIA: 25 Marks	SEE # : 75 Marks
Programme: MSC-BIOCHEMISTRY	# - Semester End Exam

No	Course Outcome (Cos): After completion of this course, the students will be able to	PSOs	Cl.Ses	BLOOM'S TAXONOMY LEVEL
CO1	Illustrate and compute the structure and function of the cell organelles.	PS01	12	AP
CO2	Demonstrate and illustrate mechanisms of cell-cell interaction.	PS02	12	AP
CO3	Categorize the concept of DNA replication process.	PS03	12	AN
CO4	Examine and compare prokaryotic and eukaryotic transcription mechanism.	PS04	12	AN
CO5	Validate the concept of translation, and post translation modifications.	PS03	12	E

UNIT-I (Lecture Hours: 12)

STRUCTURE AND FUNCTION OF CELL

Cell –(Structure of a typical cell, Differences between prokaryotic and eukaryotic cells).

Cell Organelles – (Plasma membrane, nucleus, mitochondria, Golgi bodies, lysosomes, endoplasmic reticulum, peroxisomes, plastids, vacuoles, chloroplast)

Structure and organization of membrane – (structure of fluid mosaic model, membrane lipids, membrane proteins, and membrane fluidity)

Membrane Transport –(Diffusion (passive and facilitated).Symport, antiport and uniport systems.Active transport – Na⁺, K⁺ ATPase, Ca²⁺ ATPase, secondary active transport- group translocation of glucose, ions selective channels)

Internalization of macromolecules – (Endocytosis, pinocytosis, phagocytosis and exocytosis. Receptor mediated endocytosis)

Activity

1. Make a model on Prokaryotic and Eukaryotic cell
2. Demonstration of membrane transport

UNIT-II (Lecture Hours: 12)

CELL DIVISION & CELL SIGNALLING

Cytoskeleton- (Microtubules, Micro filaments and intermediary filaments - structure and functions. Cell junctions – types, cell adhesion molecules (cadherins and selectins) cell matrix adhesion (integrin's and Desmosome) ECM components – collagen and fibronectin).

Cell Division – (Phases of eukaryotic cell cycle, check points – cyclins, maturation promoting factor (MPF), Cyclin dependent kinases, growth factors, inhibition of cell cycle progression).

Cell Signaling –(Signaling mechanism, signal molecules, signal receptors, forms of inter cellular signaling. Cell surface receptors– G protein coupled receptors, Cyclic AMP, and Cyclic GMP)

Activity

1. Quiz on Cellsignalling by kahoot
2. Group Discussion on cell signaling

UNIT-III (Lecture Hours: 12)

ORGANIZATION OF GENETIC MATERIAL

Introduction – (Central Dogma, Structural organization of prokaryotic and eukaryotic chromosome)

GENOME REPLICATION

Introduction – (Definition and Types. Messelson and Stahl's semi-conservative replication model, Conservative model and dispersive model)

Enzymology of DNA replication – (DNA polymerases, Role of Topoisomerases, DNA ligases, helicases, SSB proteins primase and Telomerase).

REPLICATION IN E.coli- (Origin, initiation, replication fork movement, elongation and termination, uni directional and bidirectional replication)

RNA REPLICATION

Replication of RNA genomes – (RNA replicase and reverse transcriptase)

Activity

1. Animation on central dogma
2. Demonstration of Replication

UNIT-IV(Lecture Hours: 12)

TRANSCRIPTION IN PROKARYOTES

ranscription signals – (In prokaryotes and eukaryotes. RNA polymerases and transcription factors - Structure and function. Process of Transcription in Prokaryotes - Initiation, elongation and termination of transcription in *E. coli*. Reverse Transcription)

TRANSCRIPTION IN EUKARYOTES

(Classes of RNA - mRNA, rRNA, tRNA, snRNA, hnRNA. RNA Polymerases, Initiation, elongation and termination)

Posttranscriptional modifications.- (Alternative splicing & mRNA processing)

Activity

1. Debate on difference between Prokaryotic and Eukaryotic transcription
2. Preparation of Model of prokaryotic transcription

UNIT-V (Lecture Hours: 12)

GENETIC CODE AND TRANSLATION IN PROKARYOTES AND EUKARYOTES

Genetic Code – (Salient features, Deciphering the genetic code - Wobble hypothesis)

TRANSLATION IN PROKARYOTES

Process of Translation – (Initiation, chain elongation and termination of polypeptide synthesis in *E. coli*.)

TRANSLATION IN EUKARYOTES

Process of Translation – (Initiation, chain elongation and termination Differences - Binding of aminoacyl t - RNA, Peptide Bond Formation and Translocation)

Post-translational modifications – (Glycosylation of protein,Protein sorting and targeting)

Regulation of gene expression in prokaryotes- (operon concept (lac, tryptophan and arabinose), DNA repair and mutation).

Activity

1. Making a video on Genetic code
2. Making a PowerPoint slides on post translational modifications

REFERENCES

TEXTBOOKS

1. Lehninger Principles of Biochemistry: International Edition – 2017 by David L. Nelson and Michael Cox; Publisher: WH Freeman; 7th ed. 2017 edition
2. Advances in Cell Biology: Volume 2, 2013 by David M. Prescott
3. Cell and Molecular Biology – DeRobertis and Saunders, 8th edition (2017).
4. Molecular Biology | Edition:5 | Mc Graw Hill Higher Education, Moscow. | HEDRICK W PHILIP AND WEAVER F ROBERT,2012

REFERENCE BOOKS

1. Molecular Cell Biology– 2016 by Arnold Berk, Chris A. Kaiser, Harvey Lodish, Angelika Amon, HiddePloegh, Anthony Bretscher, Monty Krieger and Kelsey C. Martin; Publisher: WH Freeman; 8 edition.
2. Molecular Biology of the Cell – 2014 by Bruce Alberts, Alexander D. Johnson, Julian Lewis, David Morgan, Martin Raff, and Keith Roberts; Publisher: W. W. Norton & Company; 6 edition.
3. Molecular Biology of the Gene – 2017 by James D. Watson, A. Baker Tania, P. Bell Stephen, Gann Alexander, Levine Michael and Losick Richard; Publisher: Pearson Education; Seventh edition.
4. Biochemistry – 2015 by Jeremy M. Berg, LubertStryer, John L. Tymoczko and Gregory J. Gatto; Publisher: WH Freeman; 8th ed. edition.
5. Lewin's GENES XII – 2017 by Jocelyn E. Krebs, Elliott S. Goldstein and Stephen T. Kilpatrick; Publisher: Jones and Bartlett Publishers, Inc; 12th Revised edition edition.
6. The Cell: A Molecular Approach -2013 by Geoffrey M. Cooper, and Robert E. Hausman;

CourseTitle: ENZYMES AND ENZYME TECHNOLOGY(T)	Course Code :13C
Semester: I	Course Group : M- III
Teaching Scheme in Hrs (L:T:P): (4:0:0)	Credits 4
Map Code:	Total Contact Hours: 60
CIA: 25 Marks	SEE# : 75 Marks
Programme: MSC-BIOCHEMISTR	#-Semester End Exam

No	Course Outcome(Cos): After completion of this course, the students will be able to	PSOs	Cl.Ses	BLOOM'S TAXONOMY LEVEL
CO1	Organize classification of enzymes, isolation and purification of enzymes.	PS01	12	AN
CO2	Categorize the structure of coenzymes and their role in various metabolic function.	PS02	12	AN
CO3	Distinguish the mechanism of enzyme kinetics and enzyme inhibition in enzyme catalytic reaction	PS02	12	AN
CO4	Integrate the mechanisms involved in covalent modification of Glycogen Phosphorylase and Glutamine synthetase	PS03	12	AN
CO5	Illustrate the methods of immobilizations and its applications in pharmaceutical and food industry.	PS04	12	AP

UNIT-I (Lecture Hours: 12)

ENZYME-INTRODUCTION, NOMENCLATURE AND CLASSIFICATION

Introduction–(Definition, properties and history of enzyme)

Nomenclature and classification-(International Union of Biochemistry(IUB)–6 Major Enzymes and subdivision of enzymes with examples).

Intracellular localization of enzymes–(Location of enzymes in cellular and sub cellular organelles)

ISOLATION AND PURIFICATION OF ENZYMES

Methods of homogenization-(Mammalian tissue, Plant,Fungal and Bacterial cell wall homogenization)

METHODS OF SEPARATION AND PURIFICATION OF ENZYMES

Methods that depend on size or mass- (Gelfiltration Centrifugation.- Ultracentrifugation and Dialysis –Membrane centrifugation of enzymes).

Methods based on polarity- (Ion exchange chromatography,Electrophoresis and Isoelectric focusing).

UNIT OF ENZYME ACTIVITY:

Turn over number of an enzyme – (Definition and Reaction)

International Unit of enzyme:(Definition and Reaction) –**Katal**-(Definition and Reaction)

SPECIFICITY OF ENZYME ACTIVITY

Types of specificity-

Absolute specificity, Stereo specificity, Group specificity or Relative specificity and Reaction specificity with suitable examples.

Activity

- 1 .Animations illustrating on localization of enzymes in the cell
- 2 .Group discussion on the topic how“Enzymes Help Us Digest Food”

UNIT-II (Lecture Hours: 12)

COENZYME

Introduction- (Organic compounds-prosthetic groups-Integral part of active site).

Nicotinamide coenzymes-(NAD⁺ and NADP⁺)

:Flavin nucleotides- (Structure and functions FMN and

FAD+**-Coenzyme A-** (Structure and functions CoA.SH- Structure and functions)

Biotin-(Free biotin, Protein bound biotin.-Structure and functions)

Folate coenzymes-(THF-Structure and functions)

Lipoic acid – (Coenzyme in decarboxylation of pyruvate.- Structure and functions)

TPP-Thiamine Pyrophosphate- (Structure and function)

Pyridoxal phosphate-(Structure and function)

ACTIVESITE:

Introduction-(Binding sites and catalytic sites.-Characteristic features of Active sites).

Mode of action – (Lock and Key model and Induced fit theory,)

Determination of active site residues-(Identification of substrate and inhibitors action in enzyme catalytic reaction)

Activity

- 1 Model presentation for Lock and Key and Induced fit theory
- 2 Team work for making chart preparation for Coenzymes.

UNIT-III (Lecture Hours: 12)

ENZYME KINETICS:

Steady state theory – (Definition, Rate of formation of ES, Rate of dissociation and Graph, K_m - Michaelis constant and V_{max})

Michaelis Menten Derivation-(Derive the equation and significance of M.Mequation).

LBPlot, Eadie Hofstee and Hanesplot-(Graph and significance)

MECHANISM OF BISUBSTRATE REACTION:

Single displacement and Double displacement Reaction:

Random- order mechanism- (Ternary complex formation)

Ping-pong mechanism- (Non sequential mechanism)

FACTORS INFLUENCING OR AFFECTING ENZYME CATALYTIC REACTION-(Effect of pH, Temperature, Enzyme Substrate and Time–Mechanism and Graph).

ALLOSTERIC INHIBITION AND FEEDBACK INHIBITION

Allosteric Inhibition – (Introduction and mechanism- Binding action of enzyme in Allosteric site –Allosteric activator and Allosteric inhibitors).

Feedback Inhibition(-Mechanism of action-CTP and L-Isoleucine act as a feedback inhibitor for feedback inhibition).

REVERSIBLE AND IRREVERSIBLE INHIBITION:

Reversible Inhibition:(Introduction and Types-competitive, Non-competitive and Uncompetitive

inhibition)

Competitive Inhibition –(Substrate analogue mechanism Example- Malonate reaction in TCA cycle.

Allopurinol reaction in Purine catabolism-Substrate analogue drug for Gout treatment– Reaction for inhibiting the synthesis of Uric acid).

Non competitive inhibition- (Mechanism with suitable examples- Metal ions – Lead, Mercury, Cyanide and Arsenic in Non-competitive inhibition.

Role of Non-competitive inhibitor-Cyanide in Respiratory chain reaction. (ETC))

Uncompetitive Inhibition– (Mechanism with suitable example)

Activity

1. Working model presentation on Ping-Pong mechanism.
- 2-Making poster presentation on reversible inhibitors in enzyme catalytic reaction.

UNIT-IV (Lecture Hours: 12)

COVALENT MODIFICATION OF ENZYMES:

Glycogen phosphorylase- (Phosphorylation and Dephosphorylation reaction in animal muscle cells).

Glutamine synthetase- (Covalent modification mechanism in *E.coli* bacteria).

ISOENZYMES:- (Introduction and features of Isoenzymes)

Lactate Dehydrogenase- (Introduction,Types- LDH1-5and applications of LDH)

MULTI ENZYME COMPLEX:

Pyruvate Dehydrogenase–(Introduction and mechanism of action of PDH–Steps involved in PDH reaction).

PROTEOLYTIC ENZYME:

Chymotrypsin-(Mechanism of action of proteolytic enzyme- Cleavage to specific aminoacid residues).

Carboxy peptidases- (Mechanism of action of cleavage of carboxy terminal end of protein compounds.Action of Zn in carboxypeptidase).

Lysozyme- (Mechanism of action-Digestion process of bacterial cellwall).

Ribozyme- (Non protein enzyme– Mechanism)

Abzyme- Catalytic antibody–(Mechanism of action in immune system)

Activity

- 1-Quiz-Kahoot on the topic for Proteolytic enzymes.
2. Making clay model presentation for covalent modification of enzymes

UNIT- V (Lecture Hours: 12)

ENZYME

IMMOBILIZATION:

Methods of immobilization: Types:

(Adsorption, Crosslinking, Covalent Bonding, Entrapment and Encapsulation)

Advantage and application of immobilized enzymes–(Food industry and pharmaceutical aspects).

INDUSTRIAL USES OF ENZYMES:

Textile and Food industries–(Baking of bread, brewing industries. Washing powders, immobilized glutamate dehydrogenase linked to alcohol dehydrogenase).

ROLE OF ENZYME IN DIAGNOSIS AND TREATMENT

Serum Marker enzyme– (ALT, AST, LDH, CPK, ACP)

ALP, Cancer marker enzymes-

(Prostatic acid phosphatase (PAP),

Alkaline phosphatase

And Neuron-specific enolase)

Covid-19 Viral disease-

(Angiotensin Converting Enzyme-2 receptor action-ACE-2)

(Therapeutic enzymes- Asparaginase, Collagenase, Streptokinase, Urokinase, Penicillinase etc.)

ANTIOXIDANT ENZYMES: Role in free radical scavenging activity.

-SOD, Catalase, and Glutathione peroxidase

ROLE OF ENZYME IN GENETIC ENGINEERING ASPECTS:

Enzymes in rDNA technology-Future prospects-

(Restriction endonucleases, Polymerase, Ligase, Helicase, Topoisomerase for Genetic Engineering aspects).

Activity

- 1 –Debate on different methods of enzyme immobilization

2 –Making a PPT presentation on industrial application of enzymes.

Text Books:

T1 - Fundamentals of Enzymology | Edition:2 | Oxford Science Publications, Newyork. | Nicholas c. price, Lewisstevens2010

T2-Enzymes- Biochemistry, Biotechnology,

Clinicalchemistry.|Edition:3|EastwestPressPvtLtd,NewDelhi.|TREVOR PALMER.and PHILIPBONNER (2014)

Reference Books:

R1–Lehninger Principles of Biochemistry,David.L.Nelson,MichaelM.Cox.Eight edition,2021.

R2-Harper’sIllustrated Biochemistry,Victor Rodwell,David Bender,Kathleen Bothamand Peter Kennelly,Thirt y FirstEdition -May-2018.

R2-Enzyme Chemistry: Impact and Applications,3rdEdition by ColinJSucklingand Gibson Colin.L-2010

Course Title HUMAN PHYSIOLOGY,HORMONES AND SIGNALTRANSDUCTION	Course Code : 13D
Semester :I	Course Group : M- IV
Teaching Scheme in Hrs (L:T:P) : (4:0:0)	Credits : 4
Map Code: -	Total Contact Hours: 60
CIA: 25 Marks	SEE # : 75 Marks
Programme: MSC- BIOCHEMISTRY	# - Semester End Exam

No	Course Outcome (Cos): After completion of this course, the students will be able to	PSOs	Cl. Ses	BLOOM'S TAXONOMY LEVEL
CO1	Illustrate structure and functions of digestive system and blood compositions.	PS01	12	AP
CO2	Distinguish structure and functions of respiratory, circulatory and excretory system	PS02	12	AN
CO3	Demonstrate the mechanism of nerve impulse conduction and neurotransmitters	PS02	12	AP
CO4	Apply the mechanism, function of anterior and posterior pituitary hormones	PS03	12	AP
CO5	Examine physical understanding of hormonal signal recognition	PS03	12	AN

UNIT-I

(LECTURE HOURS: 12)

DIGESTIVE SYSTEM AND BLOOD SYSTEM:

Human physiology- (Homeostasis - Mechanism of homeostasis) –(Maintenance of equilibrium of ions and other body fluids) **Digestive System-** (Introduction and physiological process) Secretion of digestive juices – Mechanism of secretion of pepsinogen and HCl.(Mechanism and action) **Digestion and absorption of Carbohydrates, Proteins and Fats.** (Process of digestion and absorption in carbohydrates, proteins and fats). **Gastro- Intestinal hormones.** – (Introduction, Receptor mechanism of GI hormones)- Action of Cholecystokinin, Disease- Gastric ulcer and Gastritis – (Introduction, etiology. mechanism and treatment) . **Blood-** Composition and function- (Introduction and types of blood) - RBC- Hemoglobin- (Introduction and structure) **Disease state-** Polycythemia, Anemia (Definition, symptoms and treatment) Leucocytes- WBC -Disease state - Leukemia, leucopenia, (Definition,symptoms and treatment) **Platelets-** (Introduction, properties Adhesion, aggregation and agglutination) **Blood**

coagulation – (Introduction and Mechanism – stages and Factors involved in blood coagulation)

Disease state –(Hemophilia- Introduction, etiology and symptoms)

Activity

1. Plasma and serum - Blood - Separation of plasma & serum.
2. Microscopic observation- WBC, RBC, Platelets.

UNIT-II (LECTURE HOURS: 9)

RESPIRATORY, CIRCULATORY AND EXCRETORY SYSTEM

Respiratory system: Introduction – (Inspiration and Expiration) Transport and exchange of respiratory gases - Transport of gases - Mechanism (Transport of oxygen and carbon dioxide) and exchange of gases between lungs and blood, between blood and tissue (Partial pressure and Diffusion method).

Buffer system - Action of oxyhemoglobin (Buffer system Bicarbonate buffer)

Circulatory System –

(Introduction, Heart as a pump-(Systolic and diastolic actions) Blood pressure.- Normal value –(Systolic and Diastolic pressure)

Excretory System – (Kidney - Structure of nephron, composition)

Urine formation – (Mechanism of urine formation, Glomerular filtration, Tubular reabsorption and Tubular secretion). **Micturition-** (Introduction and Mechanism)

Activity

1. Heart and Kidney - Display the Heart Model.
2. Structure of Nephron - chart preparation.

UNIT-III (LECTURE HOURS: 12)

NERVOUS SYSTEM AND NERVE CELL SIGNALING

Nervous system: (Nerve Cell – Introduction -Neuron – Structure and properties of neuron) **Signal transduction –** (Introduction and Mechanism, Definition, signals, ligands and receptors) **Nerve impulse transmission-** (Introduction and mechanism) (Nerve cells - Synapses, Resting membrane potential, action potential, voltage gated ion-channels, impulse transmission). **Neurotransmitters –** (Introduction, properties and types). **Criteria and mode of action of NT Acetyl Choline, Serotonin and Catecholamine –** (Introduction and mechanism of action). **Role of G-proteins in vision –** (Introduction and mechanism, G-protein visual cycle).

Activity

1. Neurotransmitters - Animation on the Nerve transmission - Mechanism.
2. Chart preparation on Neurotransmitters.

UNIT-IV (LECTURE HOURS: 12)

HORMONES -ANTERIOR AND POSTERIOR LOBE OF PITUITARY HORMONES

Introduction, classification of hormones – (Hormones - Introduction, Chemical messengers and Classification). **Hormones of the hypothalamus and pituitary -** (chemical nature, Anterior and Posterior lobe of Pituitary Hormone - secretion, release and their biological functions.) **Mechanism of**

Hormone Action – (Hormone Receptors, Signal Transduction. Mechanism of Hormone Action - **Steroid hormones and glyco protein hormones** – (General principles concerning nature and mechanism of action of steroid hormones and glyco protein hormones)

Activity

- 1.Hormones - Chart Preparation - on Hormonal regulation of body mass, eating and drinking.
- 2.Group discussion onHormones, biological clocks and daily and seasonal rhythms.

UNIT-V (LECTURE HOURS: 12)

RECEPTOR ACTION- NUCLEAR RECEPTOR , STERIOD RECEPTOR , RAS RECEPTOR AND TYROSINE RECEPTOR

Signaling by nuclear receptors: (Ligands, nuclear and orphan receptors – (Structure andfunctions).**Steroid hormone receptor** – (Mechanism of action, Signaling pathway of the steroid hormone receptor). **Signal Transduction** –(Signal Transduction by G proteins, Ras and Receptor Tyrosine Kinases. (Introduction and mechanism). **Disease** – (Signal Transduction in Human Disease- -Introduction, aetiology and mechanism).

Activity

- 1.Receptor - Animation on Receptor mechanism
- 2.Model on Receptor ligand binding - signaling action.

Text Books :

T1 Medical Physiology | Edition:5 | JP | Dr.SembulingamL AND SembulingamPrema(2012)

T2 Medical Physiology | Edition:11 | Arun printers | ChatterjeeCC(2003)

Reference Books :

R1 Endocrinology | Edition:1 | MJP | Prakash S. Lohar(2005)

R2 Molecular cell biology | Edition:6 | W.H.freeman | LodishHarvey(2007)

Course Title : LAB IN COMPUTATIONAL AND MOLECULAR MODELLING (P)	Course Code : 13P
Semester : I	Course Group : MP - I
Teaching Scheme in Hrs (L:T:P) : 0:0:5	Credits : 4 Credits
Map Code : -	Total Contact Hours : 60
CIA : 25Marks	SEE : 75 Marks
Programme: MSC-BIOCHEMISTRY	

List of Experiments:

1. Understanding PubMed database
2. Analysis of protein sequence from protein database.
3. Analysis of gene sequence from nucleotide database.
4. Getting and analysis of primary protein structure.
5. Secondary structure analysis of protein.
6. Tertiary protein structure analysis using Rasmol.
7. Introduction of various bibliographic databases.
8. Getting the gene sequences by exploring and querying the nucleic acid databases.
9. Identification of Drug Target using pathways- KEGG.
10. Introduction of National Center for Biotechnology Information (NCBI).
11. Introduction of biological search engine- Entrez
12. Introduction to literature database at NCBI and querying the PUBMED central database using the ENTREZ search engine
13. BLAST and Clustal W- NCBI blast via the NCBI web server
14. A brief visit to Ensembl
15. Construction of phylogenetic trees
16. Sequence and Structural Database
17. Gene Prediction using GenMark and GenScan
18. Molecular Docking
19. Molecular binding - Crisper

Course Title : LAB IN QUANTITATIVE ANALYSIS OF BIOLOGICAL COMPOUNDS (P)	Course Code : 23P
Semester : I & II	Course Group : MP-II
Teaching Scheme in Hrs (L:T:P) : (0:0:5)	Credits : 4 Credits
Map Code : -	Total Contact Hours : 150
CIA : 25 Marks	SEE : 75 Marks
Programme: M.Sc-BIOCHEMISTRY	

List of Experiments:

QUANTITATIVE ANALYSIS

Isolation and Estimation of Starch - Anthrone method

Isolation and Estimation of Glycogen - Colorimetry method

Estimation of Protein - Lowry's method

Estimation of Phenol - Colorimetry method

Estimation of Calcium - Titrimetry method

Estimation of Chloride - Titrimetry method

Estimation of Vitamin - C - Colorimetry method

Estimation of Phosphorous - Fiske - Subbarow method

Estimation of Total Protein, Albumin: Globulin ratio

Estimation of phenol by colorimetry method

Estimation of chloride by titrimetry method

Extractions of lipids from egg

Estimation of Apolipoprotein

Estimation of Thiamine / Riboflavin from cereals by fluorimetry methods

DETERMINATION OF LIPID PROFILE

Determination of Total Cholesterol

Determination of Phospholipids

Determination of Triglycerides

Determination of HDL Cholesterol

Determination of LDL Cholesterol

SEPERATION TECHNIQUES CHROMATOGRAPHIC ANALYSIS

Paper chromatography (ascending)

Thin layer chromatography

column chromatography

ELECTROPHORESIS

Agarose gel electrophoresis

SDS gel electrophoresis

Seperation of Sub Cellular organelles by differential centrifugation and Density Gradient centrifugation

Isolation & Estimation of DNA

Isolation & Estimation of RNA

Text Books :

T1. Biochemical Methods | Edition:2 | New Age International Publishers | S. Sadasivam A. Manickam(2005)

Reference Books :

R1. Principles and Techniques of Practical Biochemistry | Edition:5 | Cambridge University Press | KeithWilsonJohn M. Walke(2001)

Course Title: LAB IN ENZYMOLOGY, PLANT BIOCHEMISTRY AND PLANT TISSUE CULTURE (P)	Course Code : 23Q
Semester : I & II	CourseGroup : MP-III
Teaching Scheme in Hrs(L:T:P) : 0:0:5	Credits : 4 Credits
Map Code : -	Total Contact Hours: 150
CIA : 25Marks	SEE : 75 Marks
Programme: M.Sc - BIOCHEMISTRY	

List of Experiments:

DETERMINATION OF ACID PHOSPHATASE ACTIVITY

Effect of pH on Acid Phosphatase activity

Effect of Temperature on Acid Phosphatase activity

Effect of Substrate concentration on Acid Phosphatase activity

ENZYMATIC ASSAYS IN LIVER AND SERUM

Assay of SGPT in Serum

Assay of SGPT in Liver

Assay of SGOT in Serum

Assay of SGOT in Liver

Determination of Serum Acid Phosphatase activity

Determination of Serum Alkaline Phosphatase activity

Determination of Catalase from plant source – Effect of pH & Temperature

DETERMINATION OF WATER CONTENT IN DIFFERENT PARTS OF PLANTS

a. Whole Leaf Method

b. Leaf Disc Method

SEPARATION OF CHLOROPHYLL PIGMENTS

- a) Separation by Paper chromatography
- b) Separation by Thin Layer chromatography

TISSUE CULTURE TECHNIQUES – PLANT TISSUE CULTURE

- a. Callus culture for different plants - Bacopa monnieri – Neer Brahmi plant
- b. Micropropagation – Protoplast fusion techniques in plants

TextBooks:

T1. Biochemical Methods | Edition: 1 | New Age International | S. Sadasivam A. Manickam (1996)

ReferenceBooks:

R1. Principles and Techniques of Practical Biochemistry | Edition: 5 | Cambridge University Press | Keith Wilson John M. Walker (2001)

