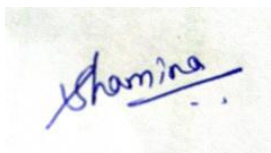


**RATHNAVEL SUBRAMANIAM COLLEGE OF ARTS & SCIENCE (AUTONOMOUS)
SULUR, COIMBATORE-641402
DEPARTMENT OF BIOCHEMISTRY
B.Sc., BIOCHEMISTRY**



**Syllabus effective for the students admitted during the academic
Year 2021 Batch & onwards**

(2021 - 2024)



HOD



PRINCIPAL



COE

PROGRAMME OUTCOMES (POs):

PO1	Graduates can have strong fundamentals in their specific discipline along with DIGITAL STRATEGIC knowledge.
PO2	To increase student's ability to communicate effectively with the community /society in verbal /written courage for such as to give or receive clear instruction.
PO3	To enhance their ability to understand and identify the professional and ethical responsibilities.
PO4	To enrich their personality and character development

PROGRAMME SPECIFIC OUTCOMES: (PSOs)

Upon completion of Bachelor of BIOCHEMISTRY Degree, STUDENTS are able to achieve the following outcomes.

PSO1	To understand the fundamental concepts and master the pertinent experimental and theoretical techniques in Molecular Gene expression, Proteomics, Pathology, Clinical research and Nano-sciences so as to inflate the understanding of biology.
PSO2	To transform the way by using sophisticated technologies and thereby gaining insights from clinical data to make cognizant decisions, predictions and to reveal the proficiency in quantitative reasoning and analytical skill within a student.
PSO3	To understand the research oriented learning that develops methodical and integrative problem-solving approaches in the biochemical industries by enabling them to write effective project reports in

	multidisciplinary environment.
PSO4	To augment the students in pre-clinical studies and cancer biology enabling them to invent new ideas to develop their entrepreneurial skills, decisive thinking and self-governance.

GRADUATE ATTRIBUTES

- DISCIPLINE KNOWLEDGE
- PROBLEM ANALYSIS
- CRITICAL THINKING
- MODERN TOOLS USAGE
- SOFT SKILLS
- SELF LEARNING
- LIFE LONG LEARNING
- INDIVIDUAL & TEAMWORK
- PROJECT MANAGEMENT & FINANCE

PROGRAMME: B.Sc., (BIOCHEMISTRY)
(Effective from the academic year 2019 - 2022)
Structure, Credits & Marks Distribution

Sl No.	Course Type	Number of Courses	Credits	Marks	Total Credits
1	Multi-Indian/ International Languages (MIL)	2	4	200	8
2	Ability Enhancement Compulsory Courses (AECC) – (I & II) : Group-I (English)	2	4	200	8
3	Ability Enhancement Compulsory Courses (AECC) –(II & IV) : Group-II	2	1+3	200	4
4	Discipline Specific Courses (DSC)	12	6	1200	72
5	Discipline Specific Elective Courses (DSE)	4+1	6	500	30
6	Extra Disciplinary Course (EDC) (DSE)	1	6	100	6
7	Skill Enhancement Courses (SEC)	2+1	4	200	8
8	ALCTA– e Learning in MOOC platform	1	4*	Pass	4*
9	Non Credit Courses – Group I	2	-	Grade	-
10	Non Credit Courses – Group II	4	-	Completed	-
Total				2600	136+4*

MULTI-INDIAN/ INTERNATIONAL LANGUAGES (MIL)								
Course	Course Name	L	T	P	CIA	ESE	Total	Credits
Two Courses – Any ONE Group								
Group I								
MIL	Tamil I	6	-	-	25	75	100	4
MIL	Tamil II	6	-	-	25	75	100	4
Group II								
MIL	Hindi I	6	-	-	25	75	100	4
MIL	Hindi II	6	-	-	25	75	100	4
Group III								
MIL	Malayalam I	6	-	-	25	75	100	4
MIL	Malayalam II	6	-	-	25	75	100	4
Group IV								
MIL	French I	6	-	-	25	75	100	4
MIL	French - II	6	-	-	25	75	100	4
Group V								
MIL	Arabic I	6	-	-	25	75	100	4
MIL	Arabic II	6	-	-	25	75	100	4
Total							200	8

ABILITY ENHANCEMENT COMPULSORY COURSES (AECC) - GROUP I : (I & II SEMESTER)								
Course	Course Name	L	T	P	CIA	ESE	Total	Credits
AECC – G-I	English I	6	-	-	25	75	100	4
AECC – G-I	English II	6	-	-	25	75	100	4
Total							200	8

ABILITY ENHANCEMENT COMPULSORY COURSES (AECC) - GROUP II : (II & IV SEMESTER)

Course	Course Name	L	T	P	CIA	ESE	Total	Credits
AECC – G-II - 1	Environmental Studies	1	-	-	100	-	100	1
AECC – G-II - 2	Aptitude	3	-	-	25	75	100	3
Total							200	4

DISCIPLINE SPECIFIC COURSES (DSC)

Course	Course Name	L	T	P	CIA	ESE	Total	Credits
DSC - I	Biomolecules	4	-	4	25	75	100	6
					40	60		
DSC - II	Allied Chemistry	4	-	4	25	75	100	6
					40	60		
DSC - III	Cellular Biochemistry	4	-	4	25	75	100	6
					40	60		
DSC - VI	Bioinstrumentation Techniques	4	-	4	25	75	100	6
					40	60		
DSC – V	Enzymology	4	-	4	25	75	100	6
					40	60		

DSC – VI	Intermediary metabolism	4	-	4	25	75	100	6
					40	60		
DSC – VII	Clinical Biochemistry	4	-	4	25	75	100	6
					40	60		
DSC –VIII	Clinical Lab Technology	4	-	4	25	75	100	6
					40	60		
DSC – IX	Molecular Biology	5	1	-	25	75	100	6
DSC – X	Immunology	4	-	4	25	75	100	6
					40	60		
DSC – XI	Biochemical Pharmacology	4	-	4	25	75	100	6
					40	60		
DSC – XII	Plant Biochemistry and Plant therapeutics	5	1	-	25	75	100	6
Total							1200	72

DISCIPLINE SPECIFIC ELECTIVE COURSES (DSE) I : (III SEMESTER)

Course	Course Name	L	T	P	CIA	ESE	Total	Credits
One Course – From the Group								
DSE - I - 1	Microbial Physiology	4	-	4	25	75	100	6
					40	60		
DSE - I - 2	Environmental Biochemistry	5	1	-	25	75	100	6
Total							100	6

DISCIPLINE SPECIFIC ELECTIVE COURSES (DSE) II : (IV SEMESTER)

Course	Course Name	L	T	P	CIA	ESE	Total	Credits
One Course – From the Group								
DSE - II - 1	Biostatistics	5	1	-	25	75	100	6
DSE - II - 2	Marine Biochemistry	5	1	-	25	75	100	6
Total							100	6

DISCIPLINE SPECIFIC ELECTIVE COURSES (DSE) III : (V SEMESTER)

Course	Course Name	L	T	P	CIA	ESE	Total	Credits
One Course – From the Group								
DSE - III - 1	Clinical Data analytics	4	-	4	25	75	100	6
					40	60		
DSE - III - 2	Stem cells and cancer biology	5	1	-	25	75	100	6
Total							100	6

DISCIPLINE SPECIFIC ELECTIVE COURSES (DSE) IV : (VI SEMESTER)								
Course	Course Name	L	T	P	CIA	ESE	Total	Credits
One Course – From the Group								
DSE –IV – 1	Human Physiology and Endocrinology	-	-	-	100	-	100	6
DSE – IV - 2	Nutritional Biochemistry	-	-	-	100	-	100	6
Total							100	6

DISCIPLINE SPECIFIC ELECTIVE COURSES (DSE) V: (V- SEMESTER)								
DSE – V - EXTRA DISCIPLINARY COURSE :(EDC)								
Course	Course Name	L	T	P	CIA	ESE	Total	Credits
One Course – From the Group								
B.A. English	Professional Communication	5	1	-	25	75	100	6
BBA	Entrepreneurship	5	1	-	25	75	100	6
BBA (CA)	Project Management	5	1	-	25	75	100	6
B.COM	Insurance and Risk Management	5	1	-	25	75	100	6
B.COM (CA)	Social Media Marketing	5	1	-	25	75	100	6
B.COM (IT)	E-Commerce	5	1	-	25	75	100	6
B.COM (PA)	Indian Tax System	5	1	-	25	75	100	6
B.COM (BA)	Digital marketing	5	1	-	25	75	100	6
BCA	Responsive Web Design	5	1	-	25	75	100	6
B.Sc., (CS)	Business Analytics	5	1	-	25	75	100	6
B.Sc., (IT)	Big Data Engineering	5	1	-	25	75	100	6
B.Sc., CSHM	Hospitality Management	5	1	-	25	75	100	6
B.Sc., ECS	Fundamentals of Digital Computers	5	1	-	25	75	100	6
B.Sc.,MATHS	Computational Mathematics	5	1	-	25	75	100	6
B.Sc., BC	Health Management	5	1	-	25	75	100	6

B.Sc., BT	Forensic Science	5	1	-	25	75	100	6
B.Sc., MB	Microbes – Health & Disease	5	1	-	25	75	100	6
B.Sc., N&D	Health & Life Style Disorders	5	1	-	25	75	100	6
Total							100	6

DISCIPLINE SPECIFIC ELECTIVE COURSES (DSE) VI : (VI SEMESTER)								
Course	Course Name	L	T	P	CIA	ESE	Total	Credits
One Course – From the group								
DSE – VI - 2	Project Report – Viva voce	-	-	-	100	-	100	6
Total							100	6

SKILL ENHANCEMENT COURSES : GROUP I (III & IV SEMESTER)								
Course	Course Name	L	T	P	CIA	ESE	Total	Credits
SEC – GI-1	Communicative Skills I	2	-	-	50	-	50	2
SEC – GI-2	Communicative Skills II	2	-	-	50	-	50	2
Total							100	4

SKILL ENHANCEMENT COURSES : GROUP II (V SEMESTER)								
Course	Course Name	L	T	P	CIA	ESE	Total	Credits
Any ONE Group								
Group A								
SEC –GII – A1	Placement - College to Corporate I	2	-	-	50	-	50	2
SEC –GII – A2	Placement - College to Corporate II	2	-	-	50	-	50	2
Group B								
SEC – GII – B	Industrial Biochemistry	4	-	-	100	-	100	4
Total							100	4

NON CREDIT COURSE – GROUP I (III & IV SEMESTER)				
NCC – G1-1 (III Semester)	Professional English - I	RVS Training Academy		Grade
NCC – G1-2 (IV Semester)	Professional English - II	RVS Training Academy		Grade

NON CREDIT COURSE – GROUP II (I - IV SEMESTER)

Any ONE Course

NCC – G II	National Service Scheme	NSS	Completion
	National Cadet Corps	NCC	Completion
	Sports	Physical Education	Completion
	Literacy & Cultural Club	Language Department	Completion
	Youth Red Cross / Red Ribbon Club	YRC	Completion
	Fine Arts Club	Language Department	Completion

EXTRA OPTIONAL CREDIT COURSE (ALCTA) I – VI SEMESTER

Any ONE Course with 4 Extra Credits

I – VI Semester	e-Learning in MOOC Platform	4 Credits	Completion
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RATHNAVEL SUBRAMANIAM COLLEGE OF ARTS & SCIENCE (Autonomous)

Sulur, Coimbatore – 641 402

SCHEME OF EXAMINATIONS

B. Sc., BIOCHEMISTRY 2021 - 2024 BATCH

Semester	Course Opted	Course Name	D	L	T	P	CIA	ESE	Marks	Credits
I	MIL - I	Tamil-I/Hindi-I / Malayalam – I/ French-I/Arabic- I	3	6	-	-	25	75	100	4
	AECC – G I -1	English-I	3	6	-	-	25	75	100	4
	DSC – I	Biomolecules	3	4	-	4	25	75	100	6
			3				40	60		
	DSC – II	Allied Chemistry	3	4	-	4	25	75	100	6
			3				40	60		

	NCC – GII	NCC/NSS/ SPORTS/CULTURALS	-	1	-	-	-	-	-	-
	LIB	Library	-	1	-	-	-	-	-	-
Total					30				400	20
II	MIL-II	Tamil-II/Hindi-II/Malayalam – II/French-II/Arabic-II	3	6	-	-	25	75	100	4
	AECC – GI -2	English-II	3	6	-	-	25	75	100	4
	DSC – III	Cellular Biochemistry	3	4	-	-	25	75	100	6
			6				40	60		
	DSC – IV	Bioinstrumentation Techniques	3	4	-	-	25	75	100	6
			6				40	60		
	AECC – G II - 1	Environmental Studies	3	1	-	-	100	-	100	1
	NCC – G II	NCC/NSS/ SPORTS/CULTURALS	-	1	-	-	-	-	-	-
LIB	Library	-	1	-	-	-	-	-	-	
Total					34				600	24

Semester	Course Opted	Course Name	D	L	T	P	CIA	ESE	Marks	Credits
III	DSC – V	Enzymology	3	4	-	4	25	75	100	6
			6				40	60		
	DSC – VI	Intermediary metabolism	3	4	-	4	25	75	100	6
			6				40	60		
	DSE - I	Elective - I	3	5	1	-	25	75	100	6
							40	60		
	SEC – G I – 1	Communicative Skills - I	3	2	-	-	50	-	50	2
	NCC – G I -1	Professional English - I	3	2	-	-	Grade			
NCC – G II	NCC/NSS/ SPORTS/CULTURALS	-	1	-	-	-	-	-	-	
Total					29				450	26
IV	DSC – VII	Clinical Biochemistry	3	4	-	4	25	75	100	6
			6				40	60		
	DSC – VIII	Clinical Lab Technology	3	4	-	4	25	75	100	6
			6				40	60		
	DSE - II	Elective - II	3	5	1	-	25	75	100	6
40							60			
AECC – G II - 2	Aptitude	3	3	-	-	25	75	100	3	

Semester	Course Opted	Course Name	D	L	T	P	CIA	ESE	Marks	Credits	
V	DSC – IX	Molecular Biology	3	4	-	4	25	75	100	6	
			6				40	60			
	DSC – X	Immunology	3	4	-	4	25	75	100	6	
			6				40	60			
	DSE-III	Elective-III	3	5	1	-	25	75	100	6	
							40	60			
	DSC – IX	Elective - V (EDC)	3	5	1	-	25	75	100	6	
	Any ONE Group										
	Group A										
	SEC – G II – A - 1	Placement - College to Corporate I	3	2	-	-	50	-	50	2	
	SEC – GII – A - 2	Placement - College to Corporate II		2	-	-	50	-	50	2	
	Group B										
	SEC – G II – B	Industrial Biochemistry	3	4	-	-	100	-	100	4	
	NCC – G II	NCC/NSS/SPORTS/CULTURALS	-	1	-	-	Good/ Satisfactory				
Total					32				500	28	
VI	DSC – XI	Biochemical Pharmacology	3	4	-	4	25	75	100	6	
			6				40	60			
	DSC – XII	Plant Biochemistry and Plant therapeutics	3	4	-	4	25	75	100	6	
			6				40	60			
	DSE-IV	DSE: Elective-IV	3	5	1	-	25	75	100	6	
	DSE - VI	Project Work & Vivo Voce	3	6	-	-	40	60	100	6	
	ALCTA *	(e-Learning in MOOC Platform)	Extra Credits							4*	
Total									200	12	
TOTAL									2600	136 +4* =140	

SEC – G I - 2	Communicative Skills - II	3	2	-	-	50	-	50	2	
NCC – G I - 2	Professional English - II	3	2	-	-	Grade				
NCC – G II	NCC/NSS/ SPORTS/CULTURALS	-	1	-	-	-	-	-	-	
Total					35				450	26

ABBREVIATIONS

MIL	- Multi Indian/ International Languages
AECC-G1	- Ability Enhancement Compulsory Courses – I & II: Group - I (English)
AECC-G2	- Ability Enhancement Compulsory Courses – II & II: Group - II
DSC	- Discipline Specific Courses
DSE	- Discipline Specific Elective Courses
EDC	- Extra Disciplinary Course
NCC	- Non Credit Course
SEC	- Skill Enhancement Courses (Group-I & II)
ALCTA	- Advanced Learners Course in Thrust Areas– e Learning in MOOC platform

DSE I - Discipline Specific Elective Courses I: (III Semester)

1. Microbial Physiology
2. Environmental Biochemistry

DSE II - Discipline Specific Elective Courses II: (IV Semester)

1. Biostatistics
2. Marine Biochemistry

DSE III- Discipline Specific Elective Courses III: (V Semester)

1. Clinical Data analytics
2. Stem cells and cancer biology

DSE IV- Discipline Specific Elective Courses IV: (VI Semester)

1. Human Physiology and Endocrinology
2. Nutritional Biochemistry

NCC - I (Non – Credit course) Group – I (Professional English)

The assessment will be done by RVS Training Academy and grade will be given based on internal evaluation in the respective semester

NCC – II (Non – Credit Course) Group – II

The students shall complete the activities in the concerned semester and completion status will be mentioned in their fifth semester mark statement. However, completing the activities listed in Group – II is mandatory to complete their degrees.

SEMESTER I

Semester	Course Opted	Course Name	D	L	T	P	CIA	ESE	Marks	Credits
I	MIL - I	Tamil-I/Hindi-I / Malayalam – I/ French-I/Arabic- I	3	6	-	-	25	75	100	4
	AECC – G I -1	English-I	3	6	-	-	25	75	100	4
	DSC – I	Biomolecules	3 3	4	-	4	25 40	75 60	100	6

	DSC – II	Allied Chemistry	3	4	-	4	25	75	100	6
			3				40	60		
	NCC – GII	NCC/NSS/ SPORTS/CULTURALS	-	1	-	-	-	-	-	-
	LIB	Library	-	1	-	-	-	-	-	-
Total					30				400	20

Course Title	: BIOMOLECULES (T)	Course Code	: 13A
Semester	: I	Course Group	: DSC-I
Teaching Scheme in Hrs (L:T:P)	: 4:0:0	Credits	: 4
Map Code	: C(THEORY CONCEPTS)	Total Contact Hours	: 60
CIA	: 25 Marks	SEE	: 75 Marks
Programme: BSC-BC			

Course outcome: (Cos)

No.	Course Outcome (Cos): After completion of this course, the students will be able to	POs & PSOs	Cl. Ses	CL
CO1	Understand the features including structure, function of Carbohydrates	PSO 1	12	U
CO2	Understand the various types and properties of lipids	PSO 2	12	U
CO3	Recognize the structure and function of amino acids	PSO 2	12	U & R
CO4	Derive the structure and the nature of the protein molecules	PSO 3	12	U & An
CO5	Explain the structure-function relationships for nucleic acids and chromatin	PSO3	6	U & An
CO6	Identify the components of nucleotide and study the structure and functions of two types of nucleic acids DNA and RNA	PSO 3	6	U & An

UNIT-1 (Lecture hours: 12)

CARBOHYDRATES THEORY

Introduction - Definition & Classification

Monosaccharides – Introduction & Classification, Stereochemistry, Cyclic structure & Anomeric forms

Haworth projection formula

Disaccharides - Introduction & Classification, chemistry, structure & functions of Sucrose, maltose & lactose

Polysaccharides - Introduction & Classification Storage polysaccharides - Starch & Glycogen (Structure & Function)

Structural polysaccharide

UNIT-2 (Lecture hours: 12)

LIPIDS THEORY

Introduction - Definition & Classification

Simple lipids - Fats, oils & waxes. Physical properties - Solubility, specific gravity, melting point, color & odor

Chemical properties of fats - Hydrolysis, Saponification Number, Iodine Number, RM value, Acid Number and Rancidity of fats

Compound lipids - Structure & functions of Phospholipids, Glycolipids and Lipoproteins

Derived lipids - Saturated, Unsaturated & Essential fatty acids.

Sterols and Steroids - Cholesterol and its structure

UNIT-3 (Lecture hours: 12)

AMINOACID THEORY

Introduction - Definition & structure

Classification - Based on functional groups, amino acids as ampholytes

Aliphatic Aminoacids - Structure and properties

Aromatic aminoacids - Structure and properties

Peptide bond - Structure & properties

Identification - N&C terminal residues

UNIT-4 (Lecture hours: 12)

PROTEIN THEORY

Introduction - Classification & properties

Structure of proteins - Primary, secondary, tertiary & Quaternary structures

Denaturation & Renaturation of proteins - Physical & chemical agents, coagulation, refolding

UNIT-5 (Lecture hours: 12)

NUCLEIC ACIDS THEORY

Introduction - Definition & Types

Purines - Structure of Adenine, Guanine, Xanthine & Hypoxanthine

Pyrimidines - Structure of Thymine, Uracil & Cytosine

Nucleosides & nucleotides - Structures & Modified Nitrogenous bases

DNA - DNA double helix - Watson & Crick model, Chargaff's rule

Types - A, B & Z forms

Denaturation & Renaturation - Hyperchromism, Effect of pH & temperature on DNA. Nucleation & zipper reaction

RNA - Types - mRNA, rRNA, tRNA, miRNA, siRNA - Structures and their biological roles

Practicals

Text Books :

Biochemistry | Edition:5 | W.H.Freeman & Company, New York | LUBERT STRYER(2015)

Essentials of Biochemistry | Edition:2 | Books and Allied (P) Ltd | U.CHAKRAPANI AND U.SATHYANARAYANA (2016)

Fundamentals of Biochemistry | Edition:II | S.Chand & Company | JAIN.J.L(2015)

Introduction to Practical Biochemistry | Edition:3 | Tata McGraw-Hill Education | David.T. Plummer AND Plummer.(2017)

Reference Books :

BIOCHEMICAL METHODS | Edition:2 | New Age International | SADASIVAM.S(2017)

Fundamentals of Biochemistry | Edition:1 | John Wiley & Sons Inc USA | DONALD VOET AND JUDITH G. VOET(2014)

LABORATORY MANUAL IN BIOCHEMISTRY | Edition:2 | NEW AGE INTERNATIONAL PUBLISHER | JAYARAMAN.J (2016)

Lehninger Principles of Biochemistry | Edition:3 | Mac millan Worth Publishers USA | DAVID.L NELSON AND MICHAEL.M.COX(2015)

Course Title	: BIOMOLECULES (P)	Course Code	: 13P
Semester	: I	Course Group	: DSC-I
Teaching Scheme in Hrs (L:T:P)	: 4:0:0	Credits	: 4
Map Code	: H(PRACTICAL EXPERIMENTS)	Total Contact Hours	: 60
CIA	: 40 Marks	SEE	: 60 Marks
Programme: BSC-BC			

List of Experiments:

Qualitative Analysis of Monosaccharides- Glucose and Fructose
 Qualitative Analysis of Disaccharides - Sucrose and Maltose
 Qualitative Analysis of Polysaccharides - Starch and Dextrin
 Determination of Acid number of edible oil.
 Determination of saponification number of edible oil
 Estimation of unsaturated fat by iodine value of oil.
 Qualitative Analysis of Non-polar, Aliphatic Aminoacids- Methionine, Leucine
 Qualitative Analysis of Aromatic Aminoacids- Tyrosine, Tryptophan
 Qualitative Analysis of Polar, Uncharged Aminoacid- Serine
 Qualitative Analysis of Charged Aminoacid- Histidine
 Identification of protein by Biuret method
 Denaturation of protein using Egg
 Isolation of protein from different organic sources
 Identification of sulfhydryl group of isolated proteins
 Quantify the amount of casein present in milk samples
 Alkali hydrolysis on RNA
 Effect of Denaturation of DNA

Course Title : ALLIED CHEMISTRY (T)	Course Code : 13B
Semester : I	Course Group : DSC
Teaching Scheme in Hrs (L:T:P) : 4:0:4	Credits : 4 Credits
Map Code : C(THEORY – APPLICATION)	Total Contact Hours : 60
CIA : 25 Marks	SEE # : 75 Marks
Programme : BSC-BC	# - Semester End Exam

No	Course Outcome (Cos): After completion of this course, the students will be able to	Pos & PSOs	Cl. Ses	BLOOM'S TAXONOMY LEVEL
CO1	Understand the basics of thermodynamic laws and principle	PO1& PS01	12	U
CO2	Demonstrate the different aspects of chemical bonding	PO1& PS01	12	An
CO3	Describe the basic rules of organic nomenclature	PO1& PS01	12	Ap
CO4	Demonstrate titrimetry experiments and assessments of important factors that could affect the analytical result.	PO1& PS01	12	Ap
CO5	Analyse the concepts of physical chemistry	PO1& PS01	6	An
CO6	Acquire knowledge on Green chemistry	PO1& PS01	6	U

UNIT-1 (LECTURE HOURS: 12)

THERMODYNAMICS THEORY

Thermodynamics - Introduction - Definition, Energy changes and Importance of Thermodynamics

Types of system - Open, Closed and Isolated

Types of process - Reversible, Irreversible and isothermal

Properties – Extensive and intensive

Laws of thermodynamics - First law and Second law. Enthalpy and Entropy

Free energy - Definition, Spontaneity, Gibbs free energy

UNIT-2 (LECTURE HOURS:12)

CHEMICAL BONDING THEORY

Structure of atoms and molecules- Introduction

Types of chemical bonds- Ionic, Covalent, and Coordinate

Hybridization - Introduction, Salient Features of Hybridization, Shapes of hybrid orbital

Structure and polarity of water – Hydrogen Bond in Water, Buffer Solutions and Preparations

UNIT-3 (LECTURE HOURS: 12)

ORGANIC CHEMISTRY THEORY

Organic chemistry - Introduction, Classification, Nomenclature

Isomers and Isomerism - Types and Molecular Formula

Stereoisomerism- Definition and Classification - Optical and Geometric isomerism

Cyclic, Acyclic and Heterocyclic compounds- Introduction, Classification, Structure of Monocyclic, Acyclic and Heterocyclic compounds

UNIT-4 (LECTURE HOURS: 12)

VOLUMETRIC METHODS THEORY

Solution - Definition of Ideal and Non-ideal Solutions, Units of Concentration, Molecular

Weight, Equivalent Weight, Molarity, Molality, Normality, PPM and Percentage solution

Concepts of volumetric analysis - Introduction, Analysis and Titrant

Principles of acid base titration - Introduction - Acids, Alkalies, Indicators and Buffers

Titration - Strong acid vs strong base, weak acid vs strong base, weak base vs strong acid

EDTA titration - Introduction - indicators for EDTA titration

UNIT-5 (LECTURE HOURS: 12)

PHYSICAL CHEMISTRY THEORY

Electrochemistry- Definition, Electrolytes, Conductance, Equivalent Conductance, Specific Conductance, Molar conductance.

Surface chemistry-Adsorption, Adsorbents, Physisorption and Chemisorption, Colloids, Gels, Emulsions, Electro Osmosis, Inhibition

Environmental chemistry - Concept and Scope of Environmental Biochemistry

Text Books:

1.B.R. Puri,L.R.Sharma and Madhan S.Pathania,Principles of physical chemistry,(2017),Vishal Publishing Co.

2.B.S. Bahl and Arun Bahl (2014) A textbook of organic chemistry ,S.Chand and Co.Ltd.

3.Principles of Inorganic chemistry | Edition:25 | Shobinlalnagin Chand & Co | Puri&Sharma(2014)

4.Text book of Organic Chemistry | Edition:28 | Sultan chand& sons | H.M. CHAWLA AND P.L. SONI (2014)

Reference Books:

1. Industrial chemistry | Edition:10 | Goel Publishing House Meerut,

India | 2.B.K. SHARMA (2007) Chemistry | Edition:4 | Houghton mifflin company new york

2. M.Satake,Y.Hayashi(2003),Colloidal and Surface chemistry,Discovery publishing house

Course Title : ALLIED CHEMISTRY ((P))	Course Code : 13Q
Semester : I	Course Group : DSC
Teaching Scheme in Hrs (L:T:P) : 4:0:4	Credits : 4 Credits
Map Code : H(PRACTICAL EXPERIMENTS)	Total Contact Hours : 120
CIA : 25 Marks	SEE # : 75 Marks
Programme : BSC-BC	# - Semester End Exam

List of Experiments:

Demonstration of Thermodynamics - Enthalpy and Entropy

Preparation of buffer solutions – phosphate buffer and citrate buffer at different pH

Estimation of hardness of water

Qualitative Analysis of Aromatic compounds - Toluene and Naphthalene

Qualitative Analysis of Aliphatic compounds – Glucose, Fructose

Qualitative Analysis of Functional group-Carboxylic acid and tryptophan

Titration for Acid-Base - Strong Acid vs Strong Base, HCl Vs NaOH

Strong base Vs weak acid, NaOH Vs CH₃COOH(acetic acid)

Weak base Vs strong acid, Ammonia (NH₃) Vs H₂SO₄

Determination of alkalinity in water

COD (chemical oxygen demand)-Measure of wastewater quality.

Course Title : BIOINSTRUMENTATION TECHNIQUES	Course Code : 23A
Semester : II	Course Group : DSC-III
Teaching Scheme in Hrs (L:T:P) : 4:0:0	Credits : 4 Credits
Map Code : C(THEORY CONCEPTS)	Total Contact Hours : 48
CIA : 25 Marks	SEE : 75 Marks
Programme: BSC-BC	

No.	<u>Course Outcome</u>	<u>POs & PSOs</u>	<u>Cl. Ses</u>	CL
CO1	Understand the terms like pH, pOH, acidic, basic, neutral solutions and universal indicator and acquire the practical skill.	PSO 1	10	U
CO2	Explore the use of suitable chromatographic or electrophoretic techniques for actual analytical problems.	PSO 2	10	Ap
CO3	Motivating the students in handling the basic apparatus and instruments.	PSO 2	10	Ap
CO4	Identifying the separation of proteins/peptides by selecting appropriate separation techniques .	PSO 2	8	R
CO5	Understand the theoretical principles of radioactivity and appreciate the uses of radioisotopes.	PSO 1	10	R

UNIT I INSTRUMENTATION TECHNIQUES THEORY (LECTURE HOURS: 24)

pH- Introduction - Definition, Determination of pH using indicators and pH meter
 Electrode - Types- calomel electrode, silver-silver chloride electrode and standard hydrogen electrode
 Henderson's Hasselbach equation - Relationship between pKa& pH
 Buffer - Buffer solutions - Acids and bases
 Buffer system - Bicarbonate Buffer system, Hb Buffer system
 INSTRUMENTATION TECHNIQUES PRACTICAL
 Determination of pH -
 Preparation of Buffers – calibration buffer 4.1, 7.0, 9.18

UNIT II

INSTRUMENTATION TECHNIQUES THEORY (LECTURE HOURS: 24)

CHROMATOGRAPHY - Definition and its types
 Paper Chromatography –Working Principle, Instrumentation & applications.
 Thin Layer chromatography - Working Principle, Instrumentation & applications
 Column chromatography -basic principle, Working Principle, Instrumentation & applications (Gas Liquid chromatography, Ion exchange chromatography, HPLC, Affinity Chromatography, Molecular sieve)
 INSTRUMENTATION TECHNIQUES PRACTICAL
 Paper Chromatography - Circular
 Thin Layer Chromatography

UNIT III (LECTURE HOURS: 24)

INSTRUMENTATION TECHNIQUES THEORY

Centrifugation - Introduction, types of centrifugation and Rotor types
 Ultra centrifugation - Working, applications and its types.
 Electrophoresis - Introduction, types and Factors affecting electrophoretic mobility

Agarose Gel Electrophoresis - Principle, technique and applications
SDS-PAGE - Principle, technique and applications
Immuno electrophoresis- Principle, technique and applications
INSTRUMENTATION TECHNIQUES PRACTICAL
Separation of sub- cellular components using centrifuge.
Separation Technique SDS-gel Electrophoresis – Demonstration

1 Horizontal and Vertical Gel Electrophoresis

2 Polyacrylamide Gel Electrophoresis

UNIT IV INSTRUMENTATION TECHNIQUES THEORY (LECTURE HOURS: 24)

COLORIMETRY - Introduction, colour and absorption spectrum

Beer-Lambert's law - Principle and laws

Working of a single cell photoelectric colorimeter - Principle instrumentation & applications

SPECTROPHOTOMETRY AND FLOURIMETRY - Principle, Instrumentation and applications.

INSTRUMENTATION TECHNIQUES PRACTICAL

Estimation of Protein by FOLIN - CIOCALTEAU method

Estimation of Urea by DAM-TSC method

Estimation of carbohydrate by colorimetric method

UNIT V INSTRUMENTATION TECHNIQUES THEORY (LECTURE HOURS: 24)

ISOTOPES AND RADIOACTIVITY - Introduction

Tracer techniques - Radioactive decay and units of radio activity - Curie, Bequerel, specific activity

Detection and measurement of radio activity - GM counter, Scintillation counting, Autoradiography

Applications of radioisotopes - Biological and Medical sciences

INSTRUMENTATION TECHNIQUES PRACTICAL

Practical: Industrial visit to learn techniques in radioactivity

Text Books :

1. Boyer, R.F. (2000). Modern Experimental Biochemistry (3rded.). Pearson Publishers, London, United Kingdom.
2. Chatwal, G.R.&Anand, S.K. (2014). Instrumental Methods of Chemical Analysis (5thed.). Himalaya Publishing House, Mumbai, India.
3. Hofmann, A. & Clokie, S. (2018). Wilson and Walker's Principles and Techniques of Biochemistry and Molecular Biology (8thed.). Cambridge University Press, Cambridge, United Kingdom. .
4. Sawhney, S.K.& Singh, R. (2005). Introductory Practical Biochemistry (2nded.). Alpha Science International Ltd. Publishers, Oxford, United Kingdom.
5. Sharma, B.K. (2011). Instrumental Methods of Chemical Analysis (1sted.). Krishna Prakashan Media Publishers, Meerut, India.
6. Veerakumari, L. (2009). Bioinstrumentation. MJP Publishers, Chennai, India
7. Wilson, K., and Walker, J., (2010). Principles and Techniques of Biochemistry and Molecular Biology, 7th Low Price Edition, Cambridge University Press, India.

Course Title	: BIOINSTRUMENTATION TECHNIQUES	Course Code	: 23P
Semester	: II	Course Group	: DSC-III
Teaching Scheme in Hrs (L:T:P)	: 0:0:4	Credits	: 2 Credits
Map Code	: H (THEORY TECHNOLOGY)	Total Contact Hours:	48
CIA	: 40 Marks	SEE	: 60 Marks
Programme: BSC-BC			

1. Determination of pH
2. Preparation of Buffers
3. Paper Chromatography – Circular
4. Thin Layer Chromatography.
5. Separation of compounds using centrifuge.
6. SDS-gel Electrophoresis – Demonstration
7. Estimation of Protein by FOLIN - CIOCALTEAU method
8. Estimation of Urea by DAM-TSC method
9. Industrial visit to learn techniques in radioactivity

Course Title : CELLULAR BIOCHEMISTRY	Course Code : 23B
Semester : II	Course Group : DSC-IV
Teaching Scheme in Hrs (L:T:P) : 4:0:0	Credits : 4 Credits
Map Code : C(THEORY CONCEPTS)	Total Contact Hours : 48
CIA : 25 Marks	SEE : 75 Marks
Programme: BSC-BC	

No.	<u>Course Outcome</u>	<u>POs & PSOs</u>	<u>Cl. Ses</u>	CL
CO1	Recall the history of cytology and draw the structure of cell organelles and locate its parts along with functions.	PSO 1	10	U
CO2	Design the model of a cell.	PSO 2	10	Ap
CO3	Distinguish the structure of prokaryotic and eukaryotic cell	PSO 2	10	Ap
CO4	Explain the organization of Genes and chromosomes morphology and its aberrations.	PSO 2	8	R
CO5	Distinguish the types and mechanism of mutations.	PSO 1	5	R
CO6	Compare and contrast the events of cell cycle and its regulation.	PSO 1 & PSO 2	5	R & Ap

UNIT-I (LECTUREHOURS:24)

Cell biology

Cell Structure (Cytoplasm- Structure- Composition), Cellular organelles (Nucleus- Mitochondria – Golgi bodies- Lysosomes- Endoplasmic reticulum- Peroxisomes- Plastids- Vacuoles- Ribosomes), Cytoskeleton (Structure and Function).

Cell membrane ((Structure and Functions- Fluid mosaic model, Unit membrane model)

Membrane Transport (Active and Passive Transport- Endocytosis and exocytosis- Pinocytosis and Phagocytosis).

1. Microscopic view of prokaryotic cells using staining techniques.- Simple staining
2. Microscopic view of prokaryotic cells using staining techniques.- Gram staining

UNIT-II (LECTURE HOURS:24)

Cell division and cell growth

Cell division (Mitosis and Meiosis, reductive division), Cell cycle (Phases of cell cycle-S phase, G phase, M phase) cell cycle regulation and, cell cycle checkpoints)

1. Mitotic cell division by using-onion root tip

UNIT III (LECTURE HOURS:24)

Cell communication and signaling

Cell cell interaction (Cell communications – electric and chemical – signaling mechanisms- cell surface receptors).

Basic aspects of intercellular communication (Intracrine, autocrine, paracrine, endocrine & neuronal, endocrine communications)

1. Extraction of protein-Total cellular protein

UNIT-IV (LECTUREHOURS:24)

Biochemical energetics

Biological oxidation (High energy compounds-Redox potential), ETC (electron transport chain-oxidative phosphorylation-inhibition of electron transport chain).

1. Separation of nucleic acid bases - By TLC
2. Preparation of cell culture media-Preparation of cell culture media

UNIT V (LECTUREHOURS:24)

Regulation of cell growth

Cell death (apoptosis-necrosis), Tumor (Types- causes and invasion – mutation – viral infection – tumor viruses- DNA and RNA viruses – lifecycle of virus and its regulation-senescence-proliferation control)

1. Cellular separation by using centrifugation technique.
2. Isolation of genomic DNA in liver samples
3. Isolation of genomic RNA in liver samples.

TextBooks:

Cell and Molecular biology, Edition 8th, Lippincott Williams and Wilkins, Philadelphia, De Robertis, E.D.P and De Robertis, E.M.F. (2010)

The Cell : A Molecular Biology, Edition 7th, Sunderland, Mass. Sinauer Associates, Inc. Cooper, G.M and Hausman, R.E. (2018)

Cell Biology | Edition:3rd edition | Himalaya Publishing House Pvt. Ltd | Powar C B(2017)

ReferenceBooks:

Cell and Molecular Biology: Concepts and Experiments, Edition 6th, John Karp, G (2019)

Essential Cell biology : Edition 6th, Garland Science, Bruce Alberts and Dennis Bray (2018)

Cell Biology | Edition: 6 | John Wiley and Sons, Inc., Hoboken, NJ Gerald Karp (2010)

Cell Biology | Edition: 5 | Universities Press (India) Pvt. Ltd.Channarayappa (2014)

Course Title	: CELLULAR BIOCHEMISTRY	Course Code	: 23Q
Semester	: II	Course Group	: DSC-IV
Teaching Scheme in Hrs (L:T:P)	: 0:0:4	Credits	: 4 Credits
Map Code	: C(THEORY CONCEPTS)	Total Contact Hours	: 48
CIA	: 25 Marks	SEE	: 75 Marks
Programme: BSC-BC			

1. Microscopic view of prokaryotic cells using staining techniques. - Simple staining
2. Microscopic view of prokaryotic cells using staining techniques - Gram staining
3. Mitotic cell division by using - onion root tip
4. Extraction of protein - Total cellular protein
5. Separation of nucleic acid bases - By paper chromatography.
6. Preparation of cell culture media - Preparation of cell culture media